

# Phosphorus(III)-Mediated Stereoconvergent Formal [4+1]-Cycloannulation of 1,2-Dicarbonyls and o-Quinone Methides: A Multicomponent Assembly of 2,3-Dihydrobenzofurans

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Supporting Information

**ABSTRACT:** A phosphorus(III)-mediated formal [4+1]-cycloaddition of 1,2-dicarbonyls and o-quinone methides to provide 2,3-dihydrobenzofurans is described. By exploiting the carbene-like nature of dioxyphospholenes, dihydrobenzofurans bearing a quaternary center at C2 are obtained in 30–92% yield with diastereoselectivities up to  $\geq$ 20:1. This study highlights the subtle steric interactions involved in the [4+1]-cycloannulation and the impact they have on yield and stereoselectivity in dihydrobenzofuran formation.

Inlike sequential fragment coupling/cyclization approaches, pericyclic and transition-metal-catalyzed cycloadditions offer convergent entry into functionalized heterocycles. Despite impressive advances in intramolecular C-H activations, [3+2]-cycloadditions are still widely used for 5membered heterocycle construction.<sup>3</sup> In contrast, [4+1]-cycloadditions and chelotropic processes are significantly underutilized in this context (Scheme 1).4 Frequently, high activation energies and competing cyclopropane/aziridine formation from carbenoids/nitrenoids limit the versatility of these strategies.<sup>5</sup> Albeit nontrivial, addressing these challenges would enable sitespecific heterocycle functionalization through the geminal substitution of a single carbon unit, such as 2,3-dihydrobenzofurans bearing a C2 quaternary center. The 2,3-dihydrobenzofuran ring system is an important motif present in a number of natural products and pharmaceuticals for use in the treatment of an array of maladies (Figure 1).6As a result, the development of new methods toward this subunit has attracted the attention of synthetic chemists for decades.

# Scheme 1. [4+1]-Approach toward Heterocycle Construction

$$\begin{array}{c} \textit{quaternary carbon assembly} \\ \overset{\text{R}^1}{\Rightarrow} : & + & & & & & \\ \overset{\text{R}^2}{\Rightarrow} : & + & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\$$

Despite their utility, few existing methods provide direct access to 2,3-substituted dihydrobenzofurans in a stereoselective fashion. Ta,8 Speculating that a formal [4+1]-cycloaddition approach would rapidly install quaternary substitution at C2 while maintaining structural flexibility at C3, we sought to apply the retrosynthetic disconnect outlined in Scheme 1 by the addition of a carbenoid derivative into an o-quinone methide (o-QM). While decomposition of  $\alpha$ -diazo carbonyls using transition-metal catalysts is perhaps the most common means



Figure 1. Biologically active dihydrobenzofuran natural products.

of accessing carbenoid reactivity, <sup>10</sup> this approach is often hampered by the light and heat sensitivity of diazo compounds and would likely suffer from competitive C–H arylations and cyclopropanations en route to the desired dihydrobenzofurans. <sup>11</sup> Inspired by recent independent studies conducted by Radosevich and He, we chose to employ a Kukhtin–Ramirez-like condensation in which the addition of a phosphine to ketone 5 bearing an  $\alpha$ -electron-withdrawing group generates an oxyphosphonium enolate 7 that exhibits carbene-like reactivity (Scheme 2). <sup>12</sup> Generation of o-QM 6 from silyl ether 4 <sup>9G,13</sup> followed by 1,4-addition of 7 sets the stage for an O-alkylation

#### Scheme 2. Proposed 2,3-Dihydrobenzofuran Synthesis

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event, in which P=O bond formation acts as a thermodynamic driving force, to provide 2,3-dihydrobenzofuran 8. 14

At the outset of this study, we were acutely aware of the synthetic challenges this strategy carried with it, including the potential for competitive *C*-alkylation of the initial conjugate addition adduct to yield cyclopropane 9. <sup>9e,15</sup> Additionally, we were concerned that the low temperature conditions required for *o*-QM generation may be incompatible with oxyphospholene formation <sup>9c,13</sup> and competitive self-condensation of ketone 5 would hinder dihydrobenzofuran formation. <sup>12b</sup> Herein, we describe the successful implementation of this [4+1]-cycloaddition strategy toward 2,3-dihydrobenzofuran construction and the development of a three-component assembly to access vicinal 2,3-substitution with a high degree of diastereoselectivity.

To evaluate our strategy outlined in Scheme 2, we examined the addition of  $\alpha$ -ketoester 5a to o-QM precursor 4a in the presence of  $P(NMe_2)_3$  and a fluoride source (Scheme 3). While TBAF led to an intractable mixture of products, we discovered that TBAT gave dihydrobenzofuran 8a in 43% yield. The yield improved to 82% using CsF and 18-c-6, and time and temperature proved critical toward minimizing o-QM dimer formation and self-condensation of 5a. Allowing the reaction to warm slowly from  $-78\,^{\circ}$ C to room temperature, following the sequential addition of fluoride and  $P(NMe_2)_3$ , effectively eliminated these side reactions. Likewise, treatment of isolable o-QM 6a to 5a and  $P(NMe_2)_3$  yielded syn-2,3-dihydrobenzofuran 8b as a single stereoisomer in 76% yield.

# Scheme 3. Initial Findings

Excited by these initial results, we next evaluated the functional group compatibility of this approach (Scheme 4). While both electron-rich and electron-poor aryl-substituted  $\alpha$ -ketoesters were tolerated, the yield of dihydrobenzofuran 8 was higher with electron-deficient arenes. This is consistent with a mechanism involving an initial nucleophilic addition of  $P(NMe_2)_3$  to the  $\alpha$ ketoester as proposed by Ramirez and co-workers. 12cc Similarly, chloride substitution on 4b gave cycloadduct 8g in 92% yield. While benzylic substitution on 4c did not hinder formation of 8h, the reaction proceeded with 1:1 diastereoselectivity. Symmetrical and unsymmetical 1,2-diketones were effective as demonstrated by the conversion of benzil 5f to 8i in 77% yield and unsymmetrical diketone 5g to methyl ketone 8j in 48% as a single regioisomer. Interestingly, employing  $\alpha$ -ketoesters bearing alkyl and vinyl  $\beta$ -substituents failed to provide the corresponding benzofuran in appreciable yields.

Unfortunately, synthesis of the secondary benzylic bromides proved low yielding and hampered the application of this method toward 2,3-substituted cycloadducts. To address this issue we were inspired by Pettus' work on hetero-Diels—Alder cycloadditions in which he showed that addition of Grignard reagents to Boc-protected salicylaldehyde derivatives initiates a 1,5-acyl migration/ $\beta$ -elimination sequence to generate the correspond-

Scheme 4. Fluoride-Mediated o-QM Generation

R1 Br CsF, 18-c-6; then P(NMe<sub>2</sub>)<sub>3</sub>
ArC(=O)EWG 5, CH<sub>2</sub>Cl<sub>2</sub>
-78 °C → rt, 4 h

8: Ar = 4-X-C<sub>6</sub>H<sub>4</sub>
4b: R¹ = Cl: R² = H
4b: R¹ = Cl: R² = H
4c: R¹ = H; R² = Me

Starting α-keto esters 5

5a: 
$$X = H$$
,  $EWG = CO_2Me$ 
5b:  $X = Cl$ ,  $EWG = CO_2Me$ 
5c:  $X = H$ ,  $EWG = CO_2Me$ 
5c:  $X = H$ ,  $EWG = CO_2Me$ 
5d:  $X = H$ ,  $EWG = CO_2Me$ 

"Conditions: CsF (0.52 mmol) and 18-c-6 (0.52 mmol) were added to 4 (0.26 mmol) in  $CH_2Cl_2$  (0.5 M) followed by 5 (0.34 mmol) and  $P(NMe_2)_3$  (0.34 mmol).

ing o-QM. <sup>9e,18</sup> Whereas the addition of MeMgBr or MeLi/MgBr<sub>2</sub>·OEt<sub>2</sub> to salicylaldehyde **10a** failed to provide dihydrobenzofuran **8h**, the addition of MeLi, followed by **5a** and P(NMe<sub>2</sub>)<sub>3</sub>, produced adduct **8h** in 92% yield (Scheme 5). Interestingly, the addition of HMPA (1 equiv) led to a decrease in the yield of **8h**, which would seem to highlight a delicate balance in metal alkoxide charge separation for controlled o-QM generation. This modification of Pettus' method enabled ready access to 2,3-disubstituted dihydrobenzofurans.

Scheme 5. Organometallic o-Quinone Methide Generation

In general, good to excellent yields of the cycloadducts were obtained from various salicylaldehydes 10 and organolithium reagents (Scheme 6). Substitution on 10 did not adversely affect the formation of dihydrobenzofurans 8k-m, as exemplified by the presence of two electron-donating groups in 8m. Using PhLi enabled smooth o-QM formation to give adducts 8n-p. Consistent with our previous results, electron-poor aryl  $\alpha$ -ketoesters gave higher yields of 8 than their electron-rich counterparts (i.e., 8o and 8p). Employing phenyl acetylide gave 8q bearing an alkyne at C3. It is also noteworthy that although the C3-methyl-substituted dihydrobenzofuran 8r was obtained in 73% yield, employing PhLi led to a diminished yield of 8s.

Over the course of these studies, we became acutely aware of the discrepancy between the diastereoselectivities observed for Organic Letters Letter

#### Scheme 6. Dihydrobenzofuran Assembly<sup>a</sup>

$$\begin{array}{c} \text{R}^{1} & \begin{array}{c} \text{CHO} \\ \text{OBoc} \end{array} & \begin{array}{c} \text{R}^{2}\text{Li; then ArC}(=\text{O})\text{CO}_{2}\text{Me 5} \\ \hline P(\text{NMe}_{2})_{3}, \text{PhMe, } -78 \text{ $^{\circ}\text{C} \rightarrow \text{rt}} \end{array} & \begin{array}{c} \text{R}^{2} \\ \text{CO}_{2}\text{Me} \end{array} \\ \begin{array}{c} \text{5a: Ar = Ph; 5h: Ar = 3,5-(F_{3}\text{C})}_{2}\text{C}_{6}\text{H}_{3};} \end{array} & \begin{array}{c} \text{R}^{2} \\ \text{CO}_{2}\text{Me} \end{array} \end{array}$$

Starting salicylaldehyde derivatives 10

2,2,3-trisubstituted dihydrobenzofurans 8k-s

<sup>a</sup>Conditions:  $R^2Li$  (0.29 mmol) was added to **10** (0.26 mmol) in PhMe (0.2 M), followed by **5** (0.34 mmol) and  $P(NMe_2)_3$  (0.34 mmol).

dihydrobenzofurans in Schemes 4 and 6 and the high degree of selectivity for 8b (Scheme 3). Given that the E-alkylidene 6a gave exclusively the *anti* isomer of 8b, we speculated that o-QM alkylidene integrity was critical to 2,3-diastereoselectivity. Based on this assertion, it is plausible that an unselective in situ generation of the o-QM or rapid E-to-Z isomerization mediated by  $P(NMe_2)_3$  is responsible for the low diastereoselectivities observed with those substrates in Scheme 6. This is consistent with the observation that treatment of 6a, which provided 8b as a single stereoiosomer, with  $P(NMe_2)_3$  in the absence of 5a failed to provide the corresponding Z-isomer (Scheme 7). The sharp contrast in diastereoselectivity between 8b and 8k-s suggests that the selective generation of a geometrically stable o-QM alkylidene is critical to achieving high C2–C3 diastereoselectivity.

# Scheme 7. Alkylidene Geometric Stability

To test this hypothesis further, we examined 2-hydroxynaph-thaldehyde derivatives 11 with a preference for Z-alkylidene formation in the formation of dihydronaphthylfurans 12 (Scheme 8). Addition of MeLi to naphthaldehyde 11a followed by 5a and  $P(NMe_2)_3$  provided 12a in 80% yield and  $\geq$ 20:1 stereoselectivity. Electron-withdrawing halogens at C6 of the naphthyl ring were well tolerated in the cycloaddition event

Scheme 8. Dihydronaphthylfuran Synthesis $^{a,b}$ 

 $^a$ Conditions: MeLi (0.28 mmol) was added to 11 (0.26 mmol) in PhMe (0.2 M), followed by 5 (0.34 mmol) and P(NMe $_2$ ) $_3$  (0.34 mmol).  $^b$ Ratios determined by  $^1$ H NMR (500 MHz).

leading to formation of 12b and 12c in excellent yield and consistently high diastereoselectivity. Employing  $\alpha$ -keto esters 5b and 5c likewise gave the corresponding dihydronaphthylfurans 12d and 12e exclusively in 75% and 45% respectively. Strikingly, we observed a stereoconvergent dihydrobenzofuran assembly to the syn-2,3 isomer, whether employing the presumptive Z-alkylidene from 11 or E-alkylidene 6a.

On the basis of these results, a preliminary picture arises of the stereochemical control elements in this formal [4+1]-cycloannulation. Addition of  $P(NMe_2)_3$  to 5a provides an equilibrium of oxyphospholene 7a and zwitterion 7b (Scheme 9). Ramirez

Scheme 9. Dioxyphospholene/Zwitterion Formation

$$\begin{array}{c|c} O & P(NMe_2)_3 \\ \hline Ph & O \\ \hline Sa & PL_3 \\ \hline \hline Ph & O \\ \hline PL_3 \\ \hline \hline Ph & O \\ \hline$$

has shown that this equilibrium favors the more nucleophilic zwitterion 7b (L = NMe<sub>2</sub>).<sup>20</sup> If we assume an electrostatic attraction between the o-QM carbonyl oxygen and phosphonium cation, transition states 13a and 13b arise from the addition of 7b to o-QMs 6a and 11, respectively (Scheme 10). Minimizing gauche interactions orients the larger aryl group away from the o-QM benzenoid ring in 13a, resulting in phosphine oxide displacement through anti-conformer 14a to give syn-8b. Preference for syn-12 from the Z-naphthyl alkylidene is potentially due to a combination of favorable  $\pi$ - $\pi$  stacking between the aryl group on 5 and the naphthyl ring and

Scheme 10. Proposed Rationale for Diastereoselection

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minimization of Me/Ph gauche interactions in 13b. Ultimately, this leads to the same relative *syn*-C2/C3 stereochemistry and relies on configurational stability of the alkylidene to provide high levels of stereocontrol. While speculative, this hypothesis is consistent with Ramirez's findings and rationalizes the stereoselective formation of 8b and dihydronaphthylfurans.

In summary, we have developed an efficient and convergent [4+1]-cycloaddition approach toward the construction of substituted 2,3-dihydrobenzofurans. The flexibility of this strategy permits rapid access to a variety of structurally distinct dihydrobenzofurans bearing a quaternary center at the C2 position. Mechanistic studies and extension of this [4+1]-cycloaddition strategy to other heterocyclic frameworks are currently underway and will be reported in due course.

#### ASSOCIATED CONTENT

# Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b02122.

Experimental procedures and spectroscopic data for all new compounds (PDF)

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#### **Notes**

The authors declare no competing financial interest.

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# **■** REFERENCES

- (1) (a) Hummel, J. R.; Ellman, J. A. J. Am. Chem. Soc. 2015, 137, 490. (b) Liu, J.; Fang, Z.; Zhang, Q.; Liu, Q.; Bi, X. Angew. Chem. 2013, 125, 7091. (c) Jiao, L.; Yu, Z.-X. J. Org. Chem. 2013, 78, 6842. (d) Amatore, M.; Lebœuf, D.; Malacria, M.; Gandon, V.; Aubert, C. J. Am. Chem. Soc. 2013, 135, 4576. (e) Lautens, M.; Klute, W.; Tam, W. Chem. Rev. 1996, 49
- (2) (a) Mori, K.; Kurihara, K.; Yabe, S.; Yamanaka, M.; Akiyama, T. *J. Am. Chem. Soc.* **2014**, *136*, 3744. (b) Egger, J.; Carreira, E. M. *Nat. Prod. Rep.* **2014**, *31*, 449. (c) Soldi, C.; Lamb, K. N.; Squitieri, R. A.; González-López, M.; Di Maso, M. J.; Shaw, J. T. *J. Am. Chem. Soc.* **2014**, *136*, 15142.
- (3) (a) Blum, T. R.; Zhu, Y.; Nordeen, S. A.; Yoon, T. P. Angew. Chem. 2014, 126, 11236. (b) Chen, C.; Li, X.; Schreiber, S. L. J. Am. Chem. Soc. 2003, 125, 10174. (c) Longmire, J. M.; Wang, B.; Zhang, X. J. Am. Chem. Soc. 2002, 124, 13400. (d) Gothelf, A. S.; Gothelf, K. V.; Hazell, R. G.; Jørgensen, K. A. Angew. Chem., Int. Ed. 2002, 41, 4236.
- (4) (a) Wang, Y.; Muratore, M. E.; Rong, Z.; Echavarren, A. M. Angew. Chem., Int. Ed. 2014, 53, 14022. (b) Son, S.; Fu, G. C. J. Am. Chem. Soc. 2007, 129, 1046. (c) Chen, J.-R.; Hu, X.-Q.; Lu, L.-Q.; Xiao, W.-J. Chem. Rev. 2015, 115, 5301.
- (5) (a) Morandi, B.; Carreira, E. M. Science 2012, 335, 1471. (b) Künzi,
  S. A.; Morandi, B.; Carreira, E. M. Org. Lett. 2012, 14, 1900.
  (c) Denmark, S. E.; Stavenger, R. A.; Faucher, A.-M.; Edwards, J. P. J. Org. Chem. 1997, 62, 3375.
- (6) (a) Lee, I.-S.; Kim, H.-J.; Youn, U.-J.; Chen, Q.-C.; Kim, J.-P.; Ha, D. T.; Ngoc, T. M.; Min, B.-S.; Lee, S.-M.; Jung, H.-J.; Na, M.-K.; Bae, K.-

- H. Helv. Chim. Acta 2010, 93, 272. (b) Shang, S.; Long, S. Chem. Nat. Compd. 2008, 44, 186. (c) Duan, J.-a.; Wang, L.; Qian, S.; Su, S.; Tang, Y. Arch. Pharmacal Res. 2008, 31, 965. (d) Kende, A. S.; Deng, W.-P.; Zhong, M.; Guo, X.-C. Org. Lett. 2003, 5, 1785. (e) Lu King, M.; Chiang, C.-C.; Ling, H.-C.; Fujita, E.; Ochiai, M.; McPhail, A. T. J. Chem. Soc., Chem. Commun. 1982, 1150. (f) Grisham, L. M.; Wilson, L.; Bensch, K. G. Nature 1973, 244, 294.
- (7) (a) Sheppard, T. D. J. Chem. Res. 2011, 35, 377. (b) Bertolini, F.; Pineschi, M. Org. Prep. Proced. Int. 2009, 41, 385.
- (8) (a) Wu, B.; Chen, M.-W.; Ye, Z.-S.; Yu, C.-B.; Zhou, Y.-G. Adv. Synth. Catal. 2014, 356, 383. (b) Takeda, N.; Ueda, M.; Kagehira, S.; Komei, H.; Tohnai, N.; Miyata, M.; Naito, T.; Miyata, O. Org. Lett. 2013, 15, 4382. (c) Osyanin, V. A.; Osipov, D. V.; Klimochkin, Y. N. J. Org. Chem. 2013, 78, 5505. (d) Chen, M.-W.; Cao, L.-L.; Ye, Z.-S.; Jiang, G.-F.; Zhou, Y.-G. Chem. Commun. 2013, 49, 1660. (e) Jiang, H.; Sugiyama, T.; Hamajima, A.; Hamada, Y. Adv. Synth. Catal. 2011, 353, 155. (f) Körner, C.; Starkov, P.; Sheppard, T. D. J. Am. Chem. Soc. 2010, 132, 5968
- (9) (a) Osyanin, V. A.; Osipov, D. V.; Demidov, M. R.; Klimochkin, Y. N. J. Org. Chem. 2014, 79, 1192. (b) Bai, W.-J.; David, J. G.; Feng, Z.-G.; Weaver, M. G.; Wu, K.-L.; Pettus, T. R. R. Acc. Chem. Res. 2014, 47, 3655. (c) Willis, N. J.; Bray, C. D. Chem. Eur. J. 2012, 18, 9160. (d) Pathak, T. P.; Sigman, M. S. J. Org. Chem. 2011, 76, 9210. (e) Van De Water, R. W.; Pettus, T. R. R. Tetrahedron 2002, 58, 5367. (f) Yoshioka, E.; Tanaka, H.; Kohtani, S.; Miyabe, H. Org. Lett. 2013, 15, 3938.
- (10) (a) Qin, C.; Davies, H. M. L. J. Am. Chem. Soc. 2014, 136, 9792.
  (b) Ye, T.; McKervey, M. A. Chem. Rev. 1994, 94, 1091.
- (11) Miller, E. J.; Zhao, W.; Herr, J. D.; Radosevich, A. T. *Angew. Chem., Int. Ed.* **2012**, *51*, 10605.
- (12) (a) McCarthy, S. M.; Lin, Y.-C.; Devarajan, D.; Chang, J. W.; Yennawar, H. P.; Rioux, R. M.; Ess, D. H.; Radosevich, A. T. J. Am. Chem. Soc. 2014, 136, 4640. (b) Ramirez, F.; Gulati, A. S.; Smith, C. P. J. Org. Chem. 1968, 33, 13. (c) Ramirez, F. Acc. Chem. Res. 1968, 1, 168. (d) Ramirez, F.; Bhatia, S. B.; Smith, C. P. J. Am. Chem. Soc. 1967, 89, 3026. (e) Ramirez, F.; Patwardhan, A. V.; Smith, C. P. J. Org. Chem. 1966, 31, 474. (f) Zhao, W.; Fink, D. M.; Labutta, C. A.; Radosevich, A. T. Org. Lett. 2013, 15, 3090. (g) Wang, S. R.; Radosevich, A. T. Org. Lett. 2013, 15, 1926. (h) Zhou, R.; Zhang, K.; Chen, Y.; Meng, Q.; Liu, Y.; Li, R.; He, Z. Chem. Commun. 2015, 51, 14663. (i) Zhou, R.; Yang, C.; Liu, Y.; Li, R.; He, Z. J. Org. Chem. 2014, 79, 10709.
- (13) Izquierdo, J.; Orue, A.; Scheidt, K. A. J. Am. Chem. Soc. 2013, 135, 10634.
- (14) Fleury, L. M.; Wilson, E. E.; Vogt, M.; Fan, T. J.; Oliver, A. G.; Ashfeld, B. L. *Angew. Chem., Int. Ed.* **2013**, *52*, 11589.
- (15) (a) Zhang, X.-Z.; Du, J.-Y.; Deng, Y.-H.; Chu, W.-D.; Yan, X.; Yu, K.-Y.; Fan, C.-A. *J. Org. Chem.* **2016**, *81*, 2598. (b) Wilson, E. E.; Rodriguez, K. X.; Ashfeld, B. L. *Tetrahedron* **2015**, *71*, 5765.
- (16) Use of phosphines (e.g., PPh<sub>3</sub>, PnBu<sub>3</sub>, etc.) and phosphites (i.e.,  $P(OMe)_3$ ) resulted in the formation of 8a in low yields ( $\leq 10\%$ ).
- (17) Adili, A.; Tao, Z.-L.; Chen, D.-F.; Han, Z.-Y. Org. Biomol. Chem. 2015, 13, 2247.
- (18) (a) Green, J. C.; Jiménez-Alonso, S.; Brown, E. R.; Pettus, T. R. R. Org. Lett. **2011**, *13*, 5500. (b) Jones, R. M.; Selenski, C.; Pettus, T. R. R. J. Org. Chem. **2002**, *67*, 6911.
- (19) Guo, W.; Wu, B.; Zhou, X.; Chen, P.; Wang, X.; Zhou, Y.-G.; Liu, Y.; Li, C. Angew. Chem., Int. Ed. 2015, 54, 4522.
- (20) (a) Ramirez, F.; Patwardhan, A. V.; Smith, C. P. J. Org. Chem. **1965**, 30, 2575. (b) Ramirez, F.; Desai, N. B. J. Am. Chem. Soc. **1963**, 85, 3252